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measure known volume mass of water or stated volume of water into calorimeter take the temperature of the water not the mgcl 2 weigh out known mass of mgc 2 or stated mass of mgc 2 take final highest constant temperature or record temperature change rise c ho sol 641 801 160 kj mol 1

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$O_2$  reacts with water to give  $OH^-$  or equation  $O_2 + H_2O \rightarrow 2OH^-$  b apparatus  
insulated calorimeter water and thermometer measure known volume mass  
of water or  $200\text{ cm}^3$  stated volume of water into calorimeter take the  
temperature of the water not the  $MgCl_2$  weigh out known mass of  $MgCl_2$   
or stated mass of  $MgCl_2$

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1 a i positioned two surveying poles along valley floor along wadi or  
river bed one upstream 1 downstream 1 put poles vertically at 90  
degrees perpendicular 1 measured 10m between the ranging poles 10m  
apart 1 used clinometer to measure angle 1 line up position on at top  
of two poles 1 read off the angle 1 1 2 3 ii

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o2 reacts with water to give oh or equation o2 h2o 2oh b apparatus  
insulated calorimeter water and thermometer measure known volume mass  
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of water or stated volume of water into calorimeter take the temperature of the water not the  $\text{MgCl}_2$  weigh out known mass of  $\text{MgCl}_2$  or stated mass of  $\text{MgCl}_2$  take final

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$\text{O}_2$  reacts with water to give  $\text{OH}^-$  or equation  $\text{O}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{O}_2$  b apparatus insulated calorimeter water and thermometer measure known volume mass of water or  $20\text{ cm}^3$  stated volume of water into calorimeter take the temperature of the water not the  $\text{MgCl}_2$  weigh out known mass of  $\text{MgCl}_2$  or stated mass of  $\text{MgCl}_2$

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