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cannot vary in composition solutions are homogeneous at the molecular level while other mixtures are heterogeneous 3 a the process is endothermic as the solution is consuming heat

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11 a 133 cs b 127 i c 31 p 3 d 57 co 3 13 a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct

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for the correct expression $K_h = \frac{[H^+][O^{2-}]}{[H_2O]}$ $K_a = \frac{[H^+][HCOO^-]}{[HCOOH]}$ $K_b = \frac{[OH^-][H_2O]}{[HCOO^-]}$ $K_{sp} = [H_3O^+][HCOO^-]$ correct calculated concentration of H_3O^+ $[H_3O^+] = 1.2 \times 10^{-3} M$ solution manual for principles of electromagnetics 4th edition by matthew no sadiku

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