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formic acid from latin formica ant systematically named methanoic acid is the simplest carboxylic acid and has the chemical formula hoooh and structure h c o o h it is an important intermediate in chemical synthesis and occurs naturally most notably in some ants

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formic acid hco2h the simplest of the carboxylic acids used in processing textiles and leather formic acid was first isolated from certain ants and was named after the latin formica meaning ant it is made by the action of sulfuric acid upon sodium formate which is produced from carbon monoxide and sodium hydroxide

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formic acid is an irritating chemical present in the sprayed venom of some ant species and in the secretion released from some stinging nettles it s dangerous at high concentrations but at low concentrations it s very useful humans use formic acid as a food preservative since it s an antibacterial substance

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what is formic acid formic acid is a colourless fuming liquid with a pungent acrid odour with the chemical formula hcooh formic acid is systematically named as methanoic acid the common names for simple carboxylic acids come from the latin or greek names of their source

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formic acid systematically called methanoic acid is the simplest carboxylic acid it is an important intermediate in chemical synthesis and occurs naturally most famously in the venom of bee and ant stings

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foremost among insect produced chemicals formic acid is produced by ants in the subfamily formicinae almost 300 species and some bee species as a venom against predators and as a pheromone to warn fellow insects of danger

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formic acid also known as methanoic acid is the simplest form of carboxylic acid its chemical formula is hoooh which distinguishes its structure as a single carbon atom double bonded to an oxygen atom and a hydroxyl group oh

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methanoic acid better known as formic acid is the simplest carboxylic acid with the formula hcooh it is a colorless corrosive liquid with a pungent odor that is completely miscible with water and many polar solvents but only partially miscible with hydrocarbons

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formic acid or methanoic acid 1 meth əno $\bar{l}k$ hco2h a colorless corrosive liquid with a sharp odor it boils at 100 7 c and solidifies at 8 4 c it has the lowest molecular weight 2 and is the simplest of the carboxylic acids functionally it is both an acid and an aldehyde

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formic acid comes in the carboxylic acid group it is written as hoooh and formic acid is the first member of the carboxylic acid group formic acid is also known as methanoic acid to describe in detail the formic acid formula contains 2 oxygen atoms a single carbon atom and 2 hydrogen atoms

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cheap and readily available formic acid is an effective reductant for the reductive cyclization of o nitrostyrenes the reaction is air and water tolerant and provides the desired indoles in very good yields at a low catalyst loading and without generating toxic or difficult to separate byproducts

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formula ch 2 o 2 molecular weight 46 0254 iupac standard inchi 1s ch2o2 c2 1 3 h1h h 2 3 iupac standard inchikey bdagihxwwsansr uhfffaoysa n cas registry number 64 18 6 chemical structure this structure is also available as a 2d mol file or as a computed 3d sd file the 3d structure may be viewed using java or javascript

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formic acid or methanoic acid is the simplist carboxylic acid and has the chemical formula h c o 2h many animals use for defence

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formic acid fa is an organic acid with many industrial applications and the two most important applications are in silage and leather manufacturing its global production capacity was estimated to be 950 000 t y by the end of 2014 hietala et al 2016 chapter 20earth science

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